



# **THE USE OF WATER QUALITY INDEX TO EVALUATE GROUNDWATER QUALITY IN WEST OF BASRAH WELLS**

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## **ABSTRACT**

Groundwater is an important natural resource necessary for drinking use for many societies all over the world, particularly in the rural areas. Due to water shortage problem in southern part of Iraq, the need arises to use the ground water especially in west of Basrah governorate, where it is considered as an important agricultural area in Iraq that supplies other areas of the country by main vegetable crops.

Twenty-seven wells were chosen to be study cases for hydrochemical sampling and analyzing to determine the physical and chemical properties then compared with the requirements of World Health Organization (WHO 1993), and different water classifications, to estimate the groundwater suitability for different purposes.

The collected samples in the field were analyzed for total dissolved solids (TDS), electrical conductivity (EC), pH, major cations like magnesium, calcium, sodium, potassium and anions like chloride, nitrate, carbonate, bicarbonate, and sulphate, in the laboratory.

Water Quality Index (WQI) was used to assess the suitability of groundwater from the study area for human consumption. It has been calculated for each sample taken from wells and were compared to the standard guideline values as recommended by the (WHO) for drinking and public health in order to have an indication of the present quality of groundwater. According to the overall evaluation of the sites, almost all the parameters analyzed are above the desirable limits of WHO, therefore it was concluded that the quality of groundwater from the studied areas were not suitable for domestic purposes and far from drinking water standards.

**KEYWORDS:** Water quality index; Water quality; Ground water; Basrah wells

## استخدام مؤشر جودة المياه لتقييم نوعية المياه الجوفية في ابار غرب البصرة

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### الخلاصة

تعتبر المياه الجوفية من أهم الموارد الطبيعية اللازمة للشرب لكثير من المجتمعات وفي جميع أنحاء العالم، ولاسيما في المناطق الريفية. ونظرا لمشكلة نقص المياه في الجزء الجنوبي من العراق، تبرز الحاجة إلى استخدام المياه الجوفية وخاصة في غرب محافظة البصرة. حيث تعتبر منطقة زراعية مهمة في العراق والتي تزود مناطق أخرى من البلاد بمحاصيل الخضروات.

قد تم اختيار سبعة وعشرين بئرا لتكون حالات للدراسة لأخذ العينات الكيميائية وتحليلها لتحديد الخصائص الفيزيائية والكيميائية ومقارنتها مع منظمة الصحة العالمية (WHO 1993)، وتصنيفات المياه الأخرى، لتقدير مدى ملائمة المياه الجوفية لأغراض مختلفة.

وقد تم عمل تحليل مختبري لعينات المياه التي تم جمعها في المواقع كالمواد الصلبة الذائبة (TDS)، التوصيل الكهربائي (EC)، ودرجة الحموضة (PH)، الكاتيونات الرئيسية مثل المغنيسيوم والكالسيوم والصوديوم والبوتاسيوم والايونات مثل الكلوريد، النترات البيكربونات والكربونات، والكبريتات.

ثم تم حساب مؤشر جودة المياه (WQI) لتقييم مدى ملائمة المياه الجوفية في منطقة الدراسة للاستهلاك البشري ولكل نموذج مأخوذ من بئر للمياه الجوفية وتم مقارنتها بالقيم المرجعية القياسية على النحو الموصى به من قبل منظمة الصحة العالمية للشرب والصحة العامة من أجل أن يكون مؤشرا على النوعية الحالية للمياه الجوفية. وفقا لتقييم شامل من المواقع، وجد تقريبا جميع المعاملات التي تم تحليلها هي فوق الحدود المرغوب بها من منظمة الصحة العالمية، ولذلك كان الاستنتاج أن العينات غير مناسبة للأغراض المنزلية وبعيدة عن معايير مياه الشرب.

## 1. INTRODUCTION

Groundwater is an essential natural resource, depending on its usage and consumption, it is estimated that approximately one third of the world's population use groundwater for drinking, [Arumugam et al. \(2009\)](#), in recent years surface water in Iraq suffers from a decrease of flow rates due to the policy of the countries of the source, that causes a shortage water problems in Iraq for drinking and irrigation especially in Basrah governorate (south of Iraq). West of Basrah (regions of Al Zubair, Safwan, Khour Al Zubair, and Um Qaser) is far away from the rivers, and it depends mainly on ground water for irrigation in most places in that region and has been exploited to produce drinking water by using a desalination plants, therefore it is an important to study the quality of wells in west of Basrah for a particular use. Suitability of groundwater for domestic and irrigation purposes is determined by its groundwater geochemistry.

Direct rainfall recharge is considered as the main source of recharge in the study area; the Dibdibba aquifer has high values of permeabilities, direct recharge within rainstorms, accessibility and association with soils suitable for cultivation. The average of infiltration of soil is more than 2 m/day, [Manhi, \(2012\)](#).

Water Quality Index (WQI) is one of the most effective tools to communicate information on the quality of water to concerned citizens and policy makers. Therefore, it becomes an important parameter for the evaluation and management of groundwater. WQI is defined as a rating reflecting the composite influence of various water quality parameters; WQI is calculated from the standpoint of the suitability of groundwater for human consumption.

Numerous studies on Groundwater quality assessment in the West of Basrah have been made, such as [Al-Aboodi \(2003\)](#), studied groundwater characteristics in Safwan-Zubair area and found that the groundwater in studied region is not suitable for human drinking. Also, [Manhi,\(2012\)](#) studied groundwater contamination in Safwan area and concluded that the groundwater was unsuitable to direct use as a human drinking, industrial, and building purposes. On the other hand, with regarding to suitability for irrigation, the researcher concluded that the groundwater was unsuitable for irrigation, but it has been used successfully in irrigation due to high infiltration soil conditions and continuous irrigation. Accordingly, the researcher concluded that the human activities especially agriculture are the main source of the groundwater contamination together with the different waste sources such as sewage, garbage, desalination plants, and polluted air. [Mahmood, A. et al. \(2013\)](#) studied groundwater quality in Bergussia and Al -Zubair, in Basrah city,. He found the values of WQI ranged from 2052 to 3511 in Burgussia and from 2219 to 3743 in Al- Zubair and rated unfit to human consumption according to the classification of Tiwari and Mishra (1985). Moreover, [Al-Tememi, M.\(2015\)](#) studied groundwater quality and origin within Dibdibba aquifer near Jabel Sanam area southern of Basrah Governorate, Iraq, and found that the groundwater quality was classified to chloride group-sodium family-type ( $\text{Na}^+ > \text{Ca}^{2+} > \text{Mg}^{2+}$ ). On the other hand, the groundwater origin investigated by Sullin diagram which showed that old meteoric origin for all studied wells.

The aim of the present work is to discuss the suitability of groundwater for human consumption based on computed water quality index values.

## 2. AREA OF STUDY

The area of study lies between latitudes  $30^{\circ} 1' 54.3''$  N –  $30^{\circ} 29' 3.7''$  N and longitudes  $47^{\circ} 30' 52.9''$  E -  $47^{\circ} 55' 37.8''$  E , and it is a part of Dibdibba plain (the Dibdibba plain is considered as a part of southern desert extended from the south to central parts of Iraq), [Manhi, \(2012\)](#), the area of study is located in approximately 17 km west of Basrah center, The total area of study is about 2500 km<sup>2</sup>.

The Dibdiba aquifer is set by two horizontal aquifer formations. The first formation is generally unconfined and has a saturated thickness of about 20 m. The second formation makes up the rest of the aquifer thickness. A consolidated, silty, clay bed with gypsum, locally referred “Chauchab”, usually separates the two formations. The first upper portion of the aquifer is the one thought to be naturally rechargeable, and that presently plays the most significant role as a resource for groundwater-based agriculture in the Safwan, Al Zubair, and Jebal Sanam regions west of Basrah, [AL Dahaan, \(2014\)](#).

In Safwan and Um Qaser area, the upper part of the groundwater is isolated by clay layer, while in Al-Zubair area in the west side of studied area, the clay layer either absent or within the aquifer.

In the study area, due to the ground slopes towards the lowlands of Basrah center, therefore the groundwater flows towards the Shatt al Arab River.

West of Basra has a hot desert climate, like the rest of the surrounding areas, although it has a little more precipitation than the interior locations due to its location near the coast. During Summer Months, Basrah is always one of the most important cities in the world, with temperatures exceeding 50°C regularly in the months of July and August. In winter, Basra experiences mild weather with average high temperatures around 20 °C. On some Winter Nights, minimum temperatures are below 0 °C. High humidity sometimes exceeding 90% is common due to the proximity to the Arabian Gulf.

West of Basrah area is considered as an important agricultural area in Iraq that supplies other areas of the country by main vegetable crops. Recently, and in addition to some other activities, groundwater in the study area has been exploited to produce drinking water by some of desalination plants, [Manhi \(2012\)](#). The average production from these wells is 4 l/sec, the depths of wells were arranged from 20 to 30 m and encased by steel pipes that are ranged between 8 5/8 " to 13 3/4 " in diameter, [Water Resources Ministry, Water Resources directorate in Basrah, \(Unpublished report\)](#).

Based on the climate data for the meteorological station of Basra for the period 1980-2015, the climate is classified as sub-arid to arid climate due to the decrease in the annual average of rain water and an increase in the average temperature, which causes drought conditions dominate, especially after 1997. In this study and during the rainy months, the maximum monthly averages is (28.4 mm) in December, 2013, and the minimum is (3.6 mm) in October, 2014, and the total rainfall annual average is (137.4 mm), [Environmental statistics report in Iraq \(2013\)](#).

### 3. FIELD WORK

Twenty seven wells were chosen to be the sites of hydrochemical sampling, and the data were carried out by the directorate of water resources in Basra, [Water Resources Ministry, Water Resources directorate in Basrah, \(Unpublished report\)](#). The samples were taken from December, 2013 until November, 2014. [Figs. 1 and 2](#) show wells location. The locating of wells were specified by using GPS.

The water samples collected in the field were analyzed for total dissolved solids (TDS), electrical conductivity (EC), pH, major cations like magnesium, calcium, sodium, potassium and anions like chloride, nitrate, carbonate, bicarbonate, and sulphate, in the laboratory using the standard methods of American Public Health Association ([APHA 1995](#)). Samples were taken using polyethylene (PE) pre-cleaned containers. Results were evaluated according to the criteria provided by the [World Health Organization drinking water quality \(WHO 1993\)](#).

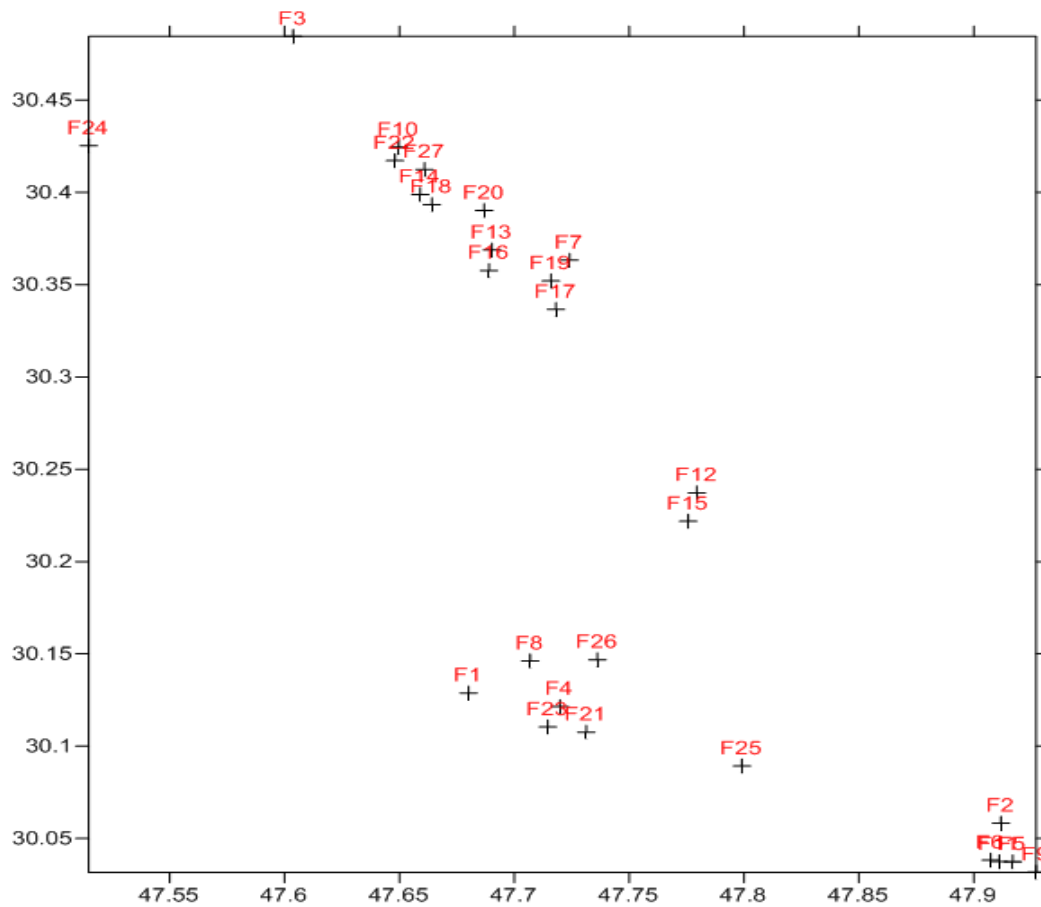


Fig. 1. Location of wells.

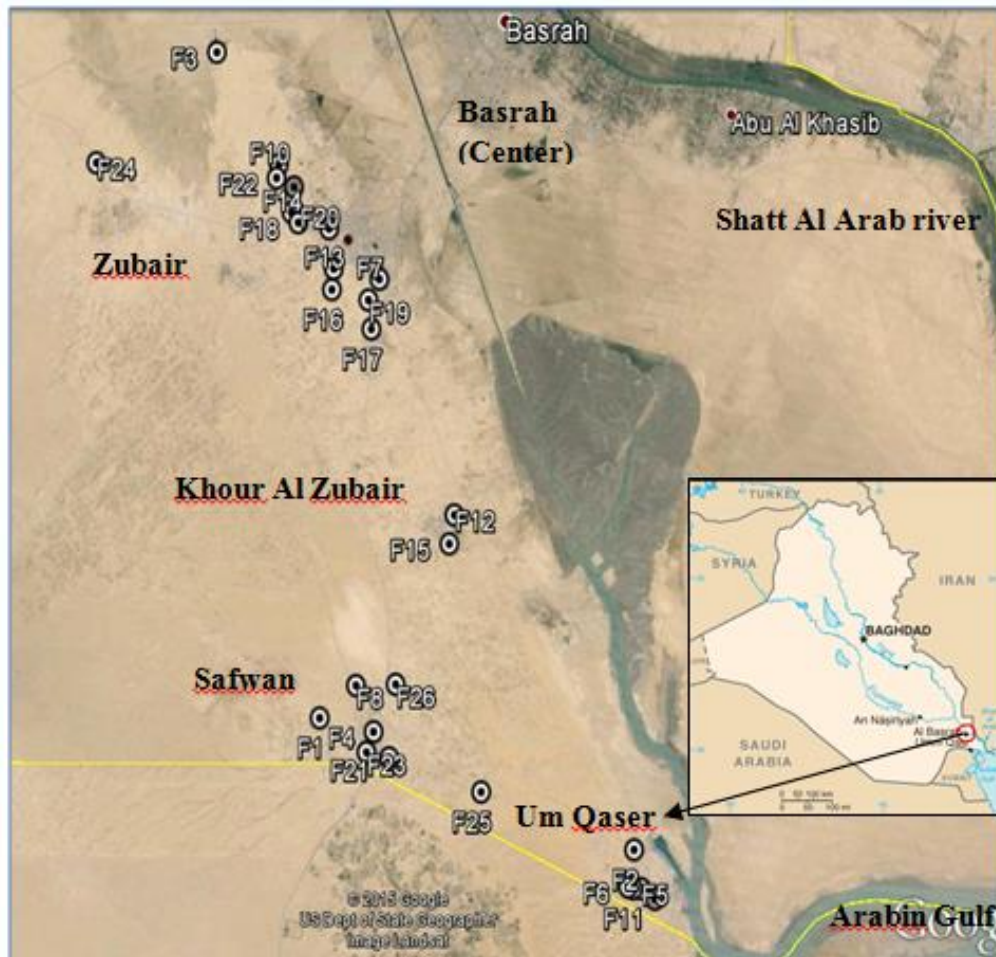


Fig. 2. Location of the study area and ground water samples.

### 3.1. DRINKING WATER QUALITY

The physical and chemical parameters of the analytical results of groundwater were compared with the standard guideline values recommended by the [World Health Organisation \(WHO 1993\)](#) for drinking and public health standards. [Table 1](#) includes the limits of the most desirable and the maximum allowable limits of various parameters.

**Table 1. Number and percentage of groundwater samples of the study area that exceeding the allowable limits prescribed by WHO for domestic purposes, Arumugam et al. (2009).**

Water quality parameters	WHO (1993)		Number of samples exceeding allowable limits	Percentage of samples exceeding allowable limits
	Most desirable limits	Maximum allowable limits		
pH	8.5	9.2	Nil	Nil
TDS (mg/l)	500	1,500	26	96.30
EC( $\mu$ S/cm)	-	1,500	27	100.00
Ca <sup>2+</sup> mg/l	75	200	27	100.00
Mg <sup>2+</sup> mg/l	50	150	16	60.00
K <sup>+</sup> mg/l	-	12	25	92.60
Na <sup>+</sup> mg/l	-	200	21	77.78
Cl <sup>-</sup> mg/l	200	600	22	81.48
NO <sub>3</sub> <sup>-</sup> mg/l	45	-	3	11.11
So <sub>4</sub> <sup>2-</sup> mg/l	200	400	26	96.30
HCO <sub>3</sub> <sup>-</sup> mg/l	-	120	19	70.30

### 3.2. CALCULATION OF WATER QUALITY INDEX (WQI)

For calculating WQI, three steps were followed. In the first step, each of the 10 parameters (TDS, pH, Cl, SO<sub>4</sub>, HCO, NO, Ca, Mg, Na, and K) has been assigned a weight (wi) based on their perceived effects on primary health (Table 2). The maximum weight of 5 has been assigned to parameters like TDS, chloride, sulfate, and nitrate due to their major importance in water quality assessment. Bicarbonate is given the minimum weight of 1 as it plays an insignificant role in the water quality assessment. Other parameters like calcium, magnesium, sodium, and potassium were assigned a weight between 1 and 5 depending on their importance in the overall quality of water for drinking purposes (Rokbani, 2011).

In the second step, the relative weight (W<sub>i</sub>) of each parameter is computed using Eq. (1):

$$W_i = \frac{w_i}{\sum_{i=1}^n w_i} \quad (1)$$

Where, w<sub>i</sub> is the weight of each parameter, n is the number of parameters and W<sub>i</sub> is the relative weight.

The weight (w<sub>i</sub>), the calculated relative weight (W<sub>i</sub>) values and the WHO standards for each parameter are given in Table 2.

In the third step, quality rating scale (q<sub>i</sub>) was calculated for each parameter using Eq. (2):

$$q_i = \frac{c_i}{s_i} * 100 \quad (2)$$



Where,  $q_i$  is the quality rating,  $C_i$  is the concentration of each chemical parameter in each water sample in mg/l and  $S_i$  is the WHO standard for each chemical parameter in mg/l (Table 2).

For computing the WQI, the  $S_{li}$  is first determined for each chemical parameter using Eq. (3) which is then used to determine the WQI as per Eq. (4).

$$S_{li} = W_i * q_i \quad (3)$$

$$WQI = \sum_{i=1}^n S_{li} \quad (4)$$

Where,  $S_{li}$  is the sub-index of  $i^{\text{th}}$  parameter  $q_i$  is the rating based on concentration of  $i^{\text{th}}$  parameter and  $n$  is the number of parameters. Computed WQI values are usually classified into five categories as shown in Table 3.

Excellent, good, poor, very poor and unfit for human consumption, Rokbani, (2011).

**Table 2. WHO standards and the calculation of relative weight (Wi)**

Chemical parameters	WHO Standard	Weight (wi)	Relative weight (Wi)
pH	8.50	3	0.103
TDS (mg/l)	500	5	0.172
Ca <sup>2+</sup> mg/l	75.0	2	0.068
Mg <sup>2+</sup> mg/l	50.0	2	0.068
K <sup>+</sup> mg/l	12.0	1	0.034
Na <sup>+</sup> mg/l	200	3	0.103
Cl <sup>-</sup> mg/l	250	3	0.103
NO <sub>3</sub> <sup>-</sup> mg/l	45.0	5	0.172
So4 <sup>2-</sup> mg/l	250	3	0.103
HCO <sub>3</sub> <sup>-</sup> mg/l	120	2	0.068
		$\sum w_i = 29$	$\sum W_i = 1$

**Table 3. Categories of WQI**

Type of water	WQI Range
Excellent water	< 50
Good water	50-100.1
Poor water	100-200.1
Very poor water	200-300.1
Unfit for drinking	> 300

#### 4. RESULTS AND DISCUSSION

Evaluating water quality index for groundwater is important to determine their suitability for drinking use. Physical and chemical parameters including statistical measures, such as minimum, maximum, average, and standard deviation of EC, TH, pH, TDS, Cl<sup>-</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, CO<sub>3</sub>, Na<sup>+</sup>, K<sup>+</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, and HCO<sub>3</sub><sup>-</sup> are listed in Table 4.



TDS amount ranges from 1200 mg/l to 10,790 mg/l with an average of 6,018 mg/l as shown in Table 4, according to WHO specification (Table 1); it can be shown that all the wells exceeds allowable limits.

The EC ranges between 1,720 and 18,030  $\mu\text{S}/\text{cm}$  with an average of 8,601  $\mu\text{S}/\text{cm}$  as shown in Table 4. Knowing that the maximum limit of EC in drinking water is prescribed as 1,500  $\mu\text{S}/\text{cm}$  at 25°C, Rokbani, (2011). From Table 4, it can be shown that all the wells were higher permitted value for drinking water purposes.

pH is one of the most important operational water quality parameters with the optimum pH at the desired range of 6.5 -8.5. The maximum allowable limit of pH in drinking water as given by the WHO is 9.2 mg/l. The values of pH in the groundwater samples collected varied from 6.80 to 8.70 with an average value of 7.48 (Table 4). This shows that the groundwater of the study area was rather alkaline in most wells, and that all the samples displayed a pH value within the maximum permissible limit.

Water with hardness (TH) higher than 200 mg/l may cause scale formation in the distribution system. The high concentration of hardness in the range of 150-300mg/l and above may cause heart diseases and kidney problems. Groundwater exceeding the limit of 300mg/l is considered to be very hard. In this study, TH is in the range of 740-6950 mg/l with an average of 2,418 mg/l, so it can be shown that all TH values exceeds the permissible limits as shown in Table 1.

The magnesium ( $\text{Mg}^{2+}$ ) ion concentration is in the range of 3.96 to 365 .49 mg/l with an average value of 165.38 mg/l. Table 4 shows that 60 % of the wells exceeds the maximum allowable limits of WHO specifications which is 150 mg/l; the high total concentrations of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are important factors which increase the hardness of waters.

Calcium ( $\text{Ca}^{2+}$ ) concentration ranges between 257 to 1,002 mg/l with an average value of 648.49 mg/l. Table 4 shows that all the wells exceeds the maximum allowable limits of WHO specifications which is 200 mg/l.

Sodium ( $\text{Na}^+$ ) concentration ranges between 12.2 to 763 mg/l with an average value of 382.11 mg/l. Table 4 shows that 78% of the wells exceeds the maximum allowable limits of WHO specifications which is 200 mg/l.

Potassium ( $\text{K}^+$ ) concentration ranges between 4.60 to 267.2 mg/l with an average value of 73.38 mg/l. Table 4 shows that all the wells except F14 and F22 exceeds the maximum allowable limits of WHO specifications which is 12 mg/l.

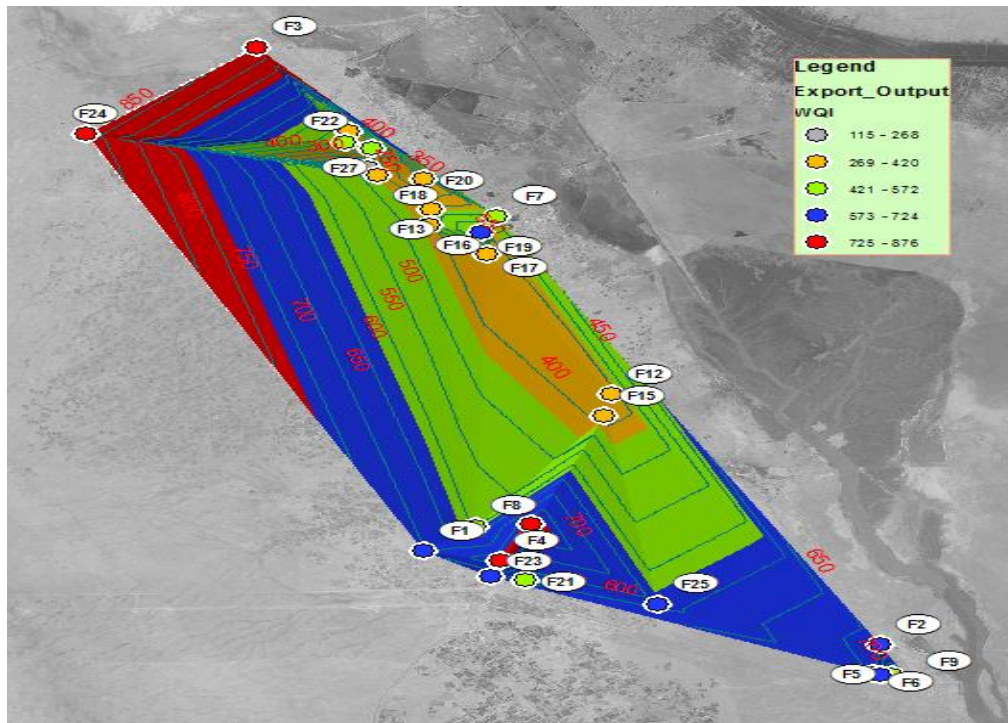
The chloride ( $\text{Cl}^-$ ) concentration ranges from 249.58 to 4,248.7 with an average value of 1,797.57 mg/l, The chloride ion concentration in groundwater of the study area exceeds the maximum permissible limit of 600mg/l in all wells excepts F9, F12, F14, F15, and F20 as shown in Table 4.

Nitrates ( $\text{NO}_3^-$ ) are the end product of aerobic stabilization of organic nitrogen and a product of conversion of nitrogenous material. This phenomenon occurs in the polluted water. Nitrate concentration of groundwater samples varied from 6 to 86 mg/l with an average value of 31.14 mg/l (Table 4). In the studied wells, nitrate concentration was below allowable limit which is (45 mg/l) in all wells except F3, F11 and F24, i.e. 89% of wells was within the acceptable limits of WHO specifications.

The bicarbonate ( $\text{HCO}_3^-$ ) ion concentration was relatively low compared to chloride, and sulphate ion concentrations range from 61 to 296 mg/l with an average value of 156.15 mg/l. Table 4 shows that 70.3% of the wells exceeds the maximum allowable limits of WHO specifications which is 120 mg/l.

From Table 4, it can be shown that west of Basrah wells were characterized by a high sulphate ion content, which is the dominant anion. Its concentration ranges between 290 and 4,408 mg/l with an average value of 2,292 mg/l. Table 4 shows that all wells except F14 exceeds the maximum allowable limits of WHO specifications, which is 400 mg/l.

The calculation of WQI is represented in Table 5. It is obvious from this classification that on the basis of the WQI groundwater from the study area is not of acceptable quality for human consumption, where all WQI values were greater than 300, Which is classified as unfit for drinking, except the sample F 14 which classified as poor water and F 20 which is classified as very poor water, See Fig. 3.



**Fig. 3 : Distribution of WQI**

**Table 4. Normal statistics of water quality parameters of groundwater samples**

Well No	Longitude	Latitude	EC $\mu\text{cm}$	TH mg/l	pH	TDS mg/l	Cl <sup>-</sup> mg/l	Mg <sup>2+</sup> mg/l	Ca <sup>2+</sup> mg/l	CO <sub>3</sub> mg/l	Na <sup>+</sup> mg/l	K <sup>+</sup> mg/l	NO <sub>3</sub> <sup>-</sup> mg/l	So <sub>4</sub> <sup>2-</sup> mg/l	HCO <sub>3</sub> <sup>-</sup> mg/l
F1	47.68011111	30.12880556	11670	2450	8.7	7650	2949	157.7	721.4	24	465.1	21	38.25	2929	122
F2	47.91183333	30.05825	11570	2750	7.5	7850	2849	279.8	641.2	24	683	202.9	39.25	2832	122
F3	47.60372222	30.48436111	18030	2850	7.7	10580	2598.8	365.49	541	60	763	65.2	52.9	4408	122
F4	47.71983333	30.12147222	11280	3150	7.7	7720	2149	291.8	781	36	596.9	242	38.6	3360	219.6
F5	47.91694444	30.0374444	9650	2750	7.6	5780	1849	316.4	581	84	503.9	34.5	28.9	2016	207.4
F6	47.90738889	30.03830556	12000	2950	7.8	7310	3049	109	1002	24	616	207	36.55	2448	97.6
F7	47.72405	30.36317222	2550	1750	7.4	5310	1499	84.6	526	24	179.5	25.7	26.55	2784	124
F8	47.70688888	30.14613889	8810	2500	7.6	5820	1749	182.1	701.4	24	527.1	121.4	29.1	2256	134.2
F9	47.92716667	30.03175	8550	2350	7.6	6140	249.58	243.34	541	160	458	57.3	7.9	2784	296
F10	47.64966667	30.4243333	4570	1900	7.2	3660	700	3.96	753.5	60	225.9	12.4	18.3	1766	122
F11	47.91108333	30.03747222	9870	2700	7.6	6990	1800	200.8	721	96	570.8	42.7	86	3408	73.2
F12	47.77944444	30.23722222	5800	6950	7.2	4350	250	210	632	84	330.7	17.9	21.75	1956	123
F13	47.69027778	30.36869444	4690	2820	7.3	3720	700	103.8	958	72	21.5	230.7	18.6	1344	292.8
F14	47.65905556	30.39877778	1720	740	7.4	1200	300	24	257	48	71.4	4.6	6	290	122
F15	47.77566667	30.22183333	5280	2200	7.3	3510	600	60	781.5	48	12.2	267.2	17.55	1448	244
F16	47.689	30.35758333	6790	1900	7.5	4730	1449	60.2	661	120	366	14.4	23.65	1776	61
F17	47.71811111	30.33652778	5950	1325	7.6	4165	850	164	260.52	39.9	127	59.6	20.825	1800	210
F18	47.66411111	30.39327778	6450	2400	7.5	5120	1400	280	501	187.2	292.3	32.7	25.6	1200	120
F19	47.71594444	30.352	12170	1700	8.1	8620	2149.3	170.3	400.8	110	251	103.8	43.1	1650	100
F20	47.68688889	30.39011111	4670	1250	7.4	3390	600	109.4	320.64	83.4	116.5	43.1	16.95	1100	120
F21	47.73130556	30.10755556	9260	2300	7.3	6710	2149	72.2	801.6	24	436.1	30.7	33.55	2496	170.8
F22	47.648	30.41711111	4280	2100	7	3960	2300	210	777.5	103.6	280	10.9	19.8	1887	134
F23	47.71458333	30.11047222	10000	1900	6.8	7150	2600	160	801.6	141	463	41.1	35.75	2100	256.2
F24	47.51469444	30.42519444	15490	2800	7.4	10790	4248.7	206.4	781.5	96	687.2	25.9	53.95	3456	292.8
F25	47.79908333	30.08927778	9970	2300	7.3	6750	2299	206.6	581.1	12	421.7	22.1	33.75	2928	122
F26	47.73613889	30.14677778	13890	2750	7.1	8990	3798	96.5	941.8	24	525	17.4	44.95	3120	97.6
F27	47.66091667	30.41233333	7290	1750	7.3	4530	1400	97	541.1	12	326.3	27	22.65	2352	109.8
Max	-	-	18,030.00	6,950.00	8.70	10,790.00	4,248.70	365.49	1,002.00	187.20	763.00	267.20	86.00	4,408.00	296.00
Min	-	-	1,720.00	740.00	6.80	1,200.00	249.58	3.96	257.00	12.00	12.20	4.60	6.00	290.00	61.00
Av.	-	-	8,601.85	2,417.96	7.48	6,018.33	1,797.57	165.38	648.49	67.45	382.11	73.38	31.14	2,292.37	156.15
SD	-	-	3,952.71	1,074.87	0.36	2,287.15	1,072.54	92.63	196.24	47.49	210.43	81.12	16.37	885.36	69.39

**Table 5. Calculation of WQI with corresponding classification**

<b>Sample</b>	<b>WQI</b>	<b>Classification</b>
F1	654.17	Unfit for drinking
F2	723.95	Unfit for drinking
F3	845.62	Unfit for drinking
F4	738.87	Unfit for drinking
F5	521.51	Unfit for drinking
F6	702.94	Unfit for drinking
F7	460.99	Unfit for drinking
F8	543.05	Unfit for drinking
F9	487.17	Unfit for drinking
F10	334.14	Unfit for drinking
F11	635.43	Unfit for drinking
F12	372.50	Unfit for drinking
F13	412.18	Unfit for drinking
F14	115.31	Poor Water
F15	389.85	Unfit for drinking
F16	408.21	Unfit for drinking
F17	350.88	Unfit for drinking
F18	416.74	Unfit for drinking
F19	586.85	Unfit for drinking
F20	271.06	Very Poor Water
F21	567.20	Unfit for drinking
F22	448.93	Unfit for drinking
F23	605.95	Unfit for drinking
F24	876.45	Unfit for drinking
F25	584.97	Unfit for drinking
F26	756.07	Unfit for drinking
F27	420.85	Unfit for drinking

#### **4.1. TOTAL DISSOLVED SOLIDS**

To establish the suitability of groundwater for any purposes, it is necessary to classify the groundwater depending upon their hydrochemical properties based on their TDS values (Freeze and Cherry 1979; Davis and DeWiest 1966) which are represented in Tables 6 and 7, respectively. The groundwater of the study area is not a fresh water it was a brackish water for all samples except F3 and F24 which was located in the north part of study area (Akl Zubair district), which is considered as a saline water based on Freeze and Cherry (1979) classification.

The study shows that there is no sample is below 1000 mg/l that mean there is no well is permissible for drinking, so there was one well which is F14 (located in the center of Al Zubair district) have a TDS value between 1000 and 3000 mg/l which is classified as useful

for irrigation and the rest of the samples classified as unfit for drinking and irrigation, because there values exceeds 3000 mg/l based on Davis and DeWiest (1966) classification.

**Table 6. Groundwater classification of all ground waters, (Freeze and Cherry, 1979)**

Total dissolved solids (mg/l)	Classification	Sample numbers	Number of samples	Percentage of samples
<1000	Fresh water type	Nil	0	0.00
1000–10,000	Brackish water type	All /F3 and F24	25	92.6
10,000–100,000	Saline water type	F3 and F24	2	7.40
>100,000	Brine water type	Nil	0	0.00
Total			27	100

**Table 7. Groundwater classification of all ground waters, (Davis and DeWiest, 1966)**

TDS (mg/l)	Classification	Sample numbers	Number of samples	Percentage of samples
<500	Desirables for drinking	Nil	0	0.00
500–1,000	Permissible for drinking	Nil	0	0.00
1,000–3,000	Useful for irrigation	F14	1	3.70
>3,000	Unfit for drinking and irrigation	All/ F14	26	96.3
Total			27	100.0

## 5. CONCLUSIONS

1. The WQI results verify that the groundwater in the study area is unsuitable for direct consumption as human drinking water.
2. The high value of WQI at these stations has been found to be mainly from the higher values of TDS in the groundwater.
3. The effect of sea water intrusion to the wells was well observed.
4. F14 well was classified as poor water, and F20 was very poor water.
5. Depending on TDS values and by using Davis and DeWiest calcification, 96.3% of wells were considered as unfit for drinking and irrigation. And based on Freeze and Cherry classification, 92.6% were considered as brackish water type and 7.4 % as saline water type.
6. Analysis revealed that the groundwater in the region needs to be some degree of treatment before consumption, and also must be protected from the risk of contamination.

## 6. SUGGESTIONS

Where most wells were unsuitable for drinking purpose, therefore the most suitable treatment technique for salinity removal is Reverse Osmosis units.

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