

## Preparation ultra fine particles from Ceramic refractory material using CO<sub>2</sub> Laser

Tarik Talib Issa, Talib Khalil Ibrahim\*

Department of Physics, College of Science, University of Baghdad, Jadiriya, Baghdad, Iraq

\* Ministry of science and Technology, Physics department

### Abstract

Samples of compact magnesia and alumina were evaporated using CO<sub>2</sub>-laser .The

Processed powders were characterized by electron microscopy and both scanning and transmission electron microscope. The results indicated that the particle size for both powders have reduced largely to 0.003 nm and 0.07 nm for MgO and Al<sub>2</sub>O<sub>3</sub>, with increasing in shape sphericity.

### Keywords

Cadmium sulphide  
Hetrojunction

### Article info

Received: June. 2009

Accepted: Sep. 2009

Published: Dec. 2009

تحضير مساحيق فائقة النعومة من مواد سيراميكية صناعية باستعمال تقانة ليزر ثاني اوكسيد الكربون

طارق طالب عيسى, طالب خليل ابراهيم\*

قسم الفيزياء - كلية العلوم - جامعة بغداد - الجادرية - بغداد - العراق

\* وزارة العلوم والتكنولوجيا - الجادرية - بغداد - العراق

### الخلاصة

حضرت مساحيق فائقة النعومة لمكبوسات المواد السيراميكية , اوكسيد المغنيسيوم و اوكسيد الالمنيوم , حضرتت باستعمال تقانة ليزر ثاني اوكسيد الكربون . درست خصائصها التركيبية بواسطة الحيويد الالكتروني والمجهر الالكتروني الماسح والنافذ. النتائج بينت الاختزال بمقدار الحجم الحبيبي للمساحيق المحضرة بحيث اصبحت بمقدار 0.003 نانو متر لمسحوق اوكسيد المغنيسيوم و لاوكسيد الالمنيوم 0.07 , مع زياده في شكل الكروييه للحبيبات

### Introduction

The standard methods of making powders involve grinding, wet chemistry and solid- state (1) reactions, which have shortcoming, such as contamination, particle agglomeration, and high energy consumption. In addition the particles produced vary in size when they are packed together and sintered, some of the spaces among them are filled in by particle

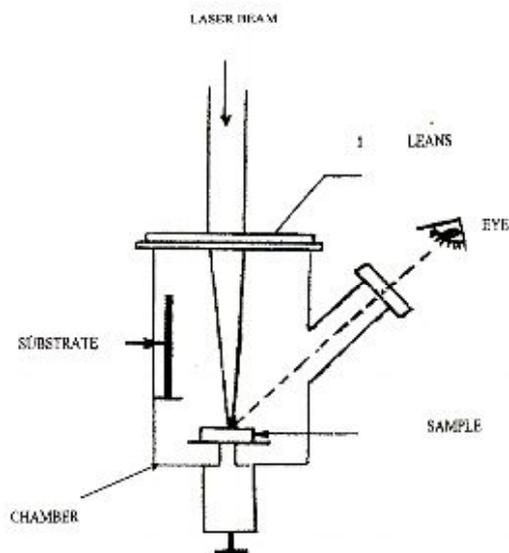
rearrangement and mass transport processes ; but voids usually remain . Laser is used in many applications, i.e. in the laboratory, in industry, and in medicine. The ideal heat source proved to process involves blowing the reaction gases through the beam of a CO<sub>2</sub> – laser to increase the reliability and decrease the resulting porosity which acts as elements of incipient cracks. At which the ceramic

could break. Researchers (2) tried to minimize the porosity before sintering using spherical particles all of one size packed in a regular manner.

### Experimental work

Two types of ceramic powders were used, MgO and Al<sub>2</sub>O<sub>3</sub> of 75µm grain size. Discs of 15mm diameter and 3mm thickness were uniaxially pressed under 3 tons force. The schematic of equipment shown in figure (1), making powders from laser heated gas. The laser beam of diameter 20mm which reflect inside the evaporation chamber by zinc- Slenoed lens of 50 mm focal length, the typical pressure of He – Ne used is (10-20) mbar, with power of 45(w).

The producing powders about 5mgm/min, and was tested by – x-ray diffraction to identify the crystal structure. Both scanning and transmission electron microscopes were used to determine both the size and the shape of the grains (sphericity).



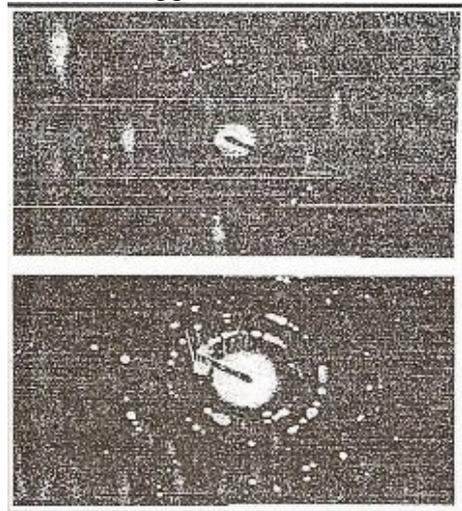
**Figure (1):** Experimental arrangement for ceramics powder processing using CO<sub>2</sub>-laser.

### Results and discussion

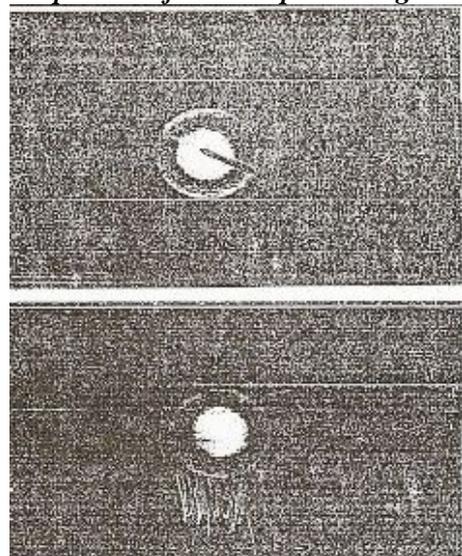
The produced powders generated by laser having either crystalline or amorphous crystal structure, with highly

reduced in its grain size and increased shape sphericity. The resulting data for MgO grain size reduced from (00095 – 0.0003) nm with crystalline form as shown in figure (2).

While for Al<sub>2</sub>O<sub>3</sub> the grain size reduced from (134.77-7) nm, which amorphous crystalain form as show in figure (3) .This is because of the rapid cooling rates leading to quench forming the amorphous structure (3).And due to laser heating which insure the formation of nucleation for all particles grain with same size (4), followed by quickly cooling, so particles that collide due to Brownim motion do not sinter to form agglomerates.



**Figure (2):**Electron diffraction of MgO powder after laser processing.



**Figure (3):**Electron diffraction of Al<sub>2</sub>O<sub>3</sub> powder after laser processing.

## Conclusions

1. Because the laser is turned to the absorption wave length of the reactants, it does not heat the container walls or other foreign substances , making the process energy efficient and producing high – purity material.
2. The reactant of  $\text{CO}_2$  – laser beam with the material to produce the ultra fine grains is controlled with constant power and low cost, comparing it with the conventional grinding techniques.
3. The produced powders was of ultra fine grain size with high sphericity .
4. low producing rate for about 5 mgm/min .
5. The production rate depends on the power of  $\text{CO}_2$  – laser and the best optimum focusing conditions.

## References

- [1]. N.W. stauffer, ed . "Moroving ceramic processing " e-lab, quantenty research bulletin of the MIT energy larb. October – December, (1980), pp.4-6.
- [2]. J.S. Hoggenty and W.R. Cannon . " Sinterable powder from laser driven reactions" . J.I. sieinfeld new York , Yoric Plenum publishing company (1981), pp.165-240.
- [3]. G.B. Kenny and H.K. Bron Ceramic Bultin 62 , 5, 590 – 596, (1983).
- [4]. M. Cauchetier , O. croix , M. luce. Advan. Ceram . Materials, (1988), 3. pp. 548.
- [5]. W.D. Callister, Material Scince and Engineering an introduction , J.W. and Sons, Inc. (2002), U.S.A.